

Now, we're marching on to CHEMISTRY!!!

Chemistry: Study of MATTER and its interaction









What is Matter?

Matter:
 • ALL MATTER HAS MASS and VOLUME
 • ALL MATTER IS MADE OF ATOMS
 • ALL MATTER HAS THE GENERAL PROPERTIES OF

Physical Property: MASS, weight, volume, density

• Describe AN object based on characteristics

Examples: • PHYSICAL PROPERTIES help identify AN object

Physical Property	How to measure the physical property	Units (if applicable)
 COLOR	Sight - colorimetric kits	% - PERCENT
MASS	 - TRIPLE BEAM BALANCE	GRAMS (g)
 VOLUME	Grad. cylinder, BEAKER, FLASK	ML
 vs.  CONDUCTIVITY		CALORIES - HEAT VHOS - ELECTRIC
 DUCTILE ABLE TO BE DRAWN INTO A THIN WIRE	RULER	MM
 MALLEABLE HAMMERED INTO THIN SHEETS	RULER	MM
 FREEZING / MELTING POINTS	THERMOMETER	°C
DENSITY	MASS / VOLUME	g/mL or g/cm ³
BUOYANCY	PLACE IN LIQUID	—