## Name:

## EOG Study Guide 2015 8<sup>TH</sup> GRADE SCIENCE

	8.P.1 - Waller: Propertie			6000			
COLOR CANNOT							
			GROUPS				
	CIIVE LUSIER						
NEGATIVELY NOCLEOS	INIE TALLUIDS	STADLE		IEIVIICAL			
1. <b>ATOMS</b> are the building blocks of m	atter.						
2. An <b>ELEMENT</b> is a pure substance the	at CANNOT be broken dow	n into a simpler substan	ce. Hint: You can fin	d them on			
the Periodic Table.							
3. A <b>COMPOUND</b> is also a pure substa	nce that is made of two or	more <b>ELEMENTS</b> that co	ombine chemically.				
4. A <b>MIXTURE</b> is two or more substance	ces that have been physica	ly combined. Each subs	tance in a mixture k	eeps its			
original <b>PROPERTIES</b>		,					
5. You can separate a mixture by FILTF	RATION, SIFTING, or EVAP	DRATION.					
6. A HETEROGENEOUS mixture is NOT	uniform throughout and t	ne components can be v	visibly distinguished.				
7. A HOMOGENEOUS mixture is unifor	rm throughout and you car	not see what it is made	of.				
8. Protons and neutrons are found in t	he NUCLEUS of an atom.						
9. Electrons are <b>NEGATIVELY</b> charged	and found in the electron	loud.					
9. The particles in a solid are <b>TIGHTLY</b>	packed. The particles in a	as are <b>NOT PACKED</b> . Th	e particles in a liqui	d are			
LOOSELY packed.							
10. The particles in a solid do not move	10. The particles in a solid do not move a lot, but <b>VIBRATE</b> in place. The particles in a gas are moving <b>VERY</b> quickly and						
freely around.							
<b>11.</b> As you move from a solid, to a liquid	d, to a gas, the particles me	ve FASTER and spread f	arther APART.				
12. The <b>PERIODIC TABLE</b> arranges elements according to increasing <b>ATOMIC NUMBER</b> .							
13. Elements in a group (family) share similar <b>PROPERTIES.</b>							
14. <b>PERIODS</b> are the horizontal rows on the periodic table. <b>GROUPS.</b> are the vertical columns on the periodic table.							
15. Non-metals are POOR conductors o	f electricity and are mostly	found on the <b>RIGHT</b> sid	e of the Periodic Ta	ble.			
16. Four properties of metals are: <b>GOOD</b> conductors of electricity, <b>DUCTILE</b> (can be made into wire), <b>MALLEABLE</b> (can be							
flattened and shaped), and have LU	<b>STER</b> (shiny).						
17. The elements along the staircase ar	e called METALLOIDS and	nave properties of meta	ls and non-metals.				
18. Groups 1, 2, and 17 are the most <b>REACTIVE,</b> group 18 (noble gases) is the most <b>STABLE.</b>							
19. A PHYSICAL change can be undone	19. A <b>PHYSICAL</b> change can be undone and nothing new is formed. For example: cutting paper, boiling water, and						
melting ice cream.							
20. A CHEMICAL change cannot be und	one and something new is	created. For example: ru	ust, burning, and rot	ting.			
21. Evidence of a chemical change inclu	21. Evidence of a chemical change includes <b>COLOR</b> change, formation of a precipitate, <b>GAS/BUBBLES</b> , and increase in						
temperature (heat given off).							
22. The LAW of CONSERVATION of MA	22. The LAW of CONSERVATION of MASS states the mass and number of atoms in the reactants is equal to the mass and						
number of atoms in the products. N	latter <b>CANNOT</b> be created	or destroyed.					
23. The <b>REACTANTS</b> are on the left side	of a chemical equation, P	RODUCTS are on the right	nt side.				

	<u>8.E.2 – Earth History</u>	<u>8.L.3 – Ecosystems</u>				
	OLDEST SUPERPOSITION PANGAEA					
CONTINENTAL DRIFT INDEX CENEZOIC		COMMENSALISM PREDATION SUN				
	ICE CORES RADIOACTIVE DATING	SYMBIOTIC HERBIVORES HETEROTROPH				
F	OSSILS WEATHERING YOUNGEST	DEPENDENT PARASITISM				
	ROCKS FOSSIL ABSOLUTE	PRODUCERS MUTUALISM AUTOTROPH				
	ICE CORES PRECAMBRIAN EROSION	INDEPENDENT LOST				
	<b>RELATIVE DEPOSITION</b>	CONSUMERS 10% PRODUCERS				
<ol> <li>1.</li> <li>2.</li> <li>3.</li> <li>4.</li> <li>5.</li> <li>6.</li> <li>7.</li> <li>8.</li> <li>9.</li> <li>10.</li> </ol>	The Theory of <b>CONTINENTAL DRIFT</b> states that the Earth's plates were once connected. The supercontinent that existed millions of years ago was called <b>PANGAEA</b> . The Law of <b>SUPERPOSITION</b> states that in any undisturbed sedimentary of rocks, the <b>YOUNGEST</b> layer is on top and the <b>OLDEST</b> is on the bottom. <b>INDEX</b> fossils existed during a specific geologic age and can be used to predict the <b>RELATIVE</b> age of a rock (an approximation). <b>ABSOLUTE</b> age provides you the exact age of a rock or fossil and is determined by using <b>RADIOACTIVE</b> <b>DATING</b> . A <b>FOSSIL</b> is the preserved remains or traces of an organism that lived in the past and can give scientists a lot of information about Earth's past. <b>ICE CORES</b> are used to study how atmospheric conditions have changed throughout Earth's history. Most of Earth's history took place in the first era, known as the <b>PRECAMBRIAN</b> Era during which there was little to no life on Earth. The era we are currently living in is known as the <b>CENEZOIC</b> Era. We use evidence from <b>ROCKS</b> , <b>FOSSILS</b> , and <b>ICE CORES</b> to show that the Earth is constantly changing. <b>WEATHERING</b> is the breaking down of rock, <b>EROSION</b> is the movement of broken sediments, and <b>DEPOSITION</b> is the building up of new rock layers.	PRODUCERS       MUTUALISM       AUTOTROPH         INDEPENDENT       LOST         CONSUMERS       10%       PRODUCERS         1.       A density DEPENDENT factor affects large populations and are often caused by overcrowding. For example: competition for food, water, shelter, space, and the quick spread of disease through a population.         2.       A density INDEPENDENT factor affect all populations regardless of their size. For example: natural disasters and human activities (pollution).         3.       PREDATION is an interaction between species in which one species eats the other.         4.       A SYMBIOTIC relationship exists between organisms of two different species that live together in direct contact.         5.       PARASITISM is a symbiotic relationships in which one organisms benefits at the expense of the other organisms benefits. For example: an oxpecker and an ox.         7.       COMMENSALISM is a symbiotic relationship in which both organisms benefits. For example: an oxpecker and an ox.         8.       The SUN is the ultimate source of the energy in an ecosystem.         9.       The first trophic level includes PRODUCERS which make their own food via photosynthesis.         10.       Another name for a consumer is a HETEROTROPH.         11.       The second trophic level includes HERBIVORES, which only only eat plants.         12.       Another name for a consumer is a HETEROTROPH.         13.       As you move up an energy pyramid, energy is LOST;				
	<u>8.L.2 – I</u>	Biotechnology				
	PRODUCTS HEALTHCARE BIOTECHNO BENEFITS ENVIRONMENT PF	LOGY LIVING AGRICULTURE RISKS COBLEMS ECONOMICALLY JOBS				
1.	Biotechnology is the use of LIVING organisms to solve P	ROBLEMS and make useful PRODUCTS For example: creating				
-	pest resistant crops and producing new medicines.					
2.	The three main areas where you see biotechnology use	are AGRICULTURE, ENVIRONMENT, and HEALTHCARE.				
კ. ⊿	<ol> <li>Some the RISKS of biotechnology are: moral concerns, safety concerns, and unknown side effects.</li> <li>Some the PENELITE of biotechnology are: finding success to discourse reducing the use of particides on event.</li> </ol>					
4.	some the <b>BENEFITS</b> of blotechnology are: finding cures	to diseases, reducing the use of pesticides on crops,				
_	increasing crop yields, and discovering new medicines.					

5. North Carolina is considered a **BIOTECHNOLOGY** state because the state has benefitted **ECONOMICALLY** from the use of biotechnology in agriculture, which has created a lot of **JOBS**.

	8.L.4 - Evolution and Genetics							
CHARLES DARWIN EVOLUTION SURVIVAL		ADAPTAT	ION BIOLOGICAL C	LASSIFICATION				
FINCHES EXTINCT		EXTINCT	ANALAGOUS	FITTEST	HOMOLOGOUS	ADAPT		
VARIATION NATURAL S			ELECTION	ENVIRONMENT				
1.	The Theory of EVOLUTION states that species change gradually over time in response to changes in their ENVIRONMENT.							
<ol> <li>2.</li> <li>3.</li> <li>4.</li> <li>5.</li> <li>6.</li> <li>7.</li> <li>8.</li> </ol>	<ol> <li>When an organism's environment changes they must ADAPT or they will become EXTINCT.</li> <li>Genetic VARIATION increases a species chance for survival.</li> <li>The ability of organisms to pass on favorable traits to their offspring is called SURVIVAL of the FITTEST.</li> <li>A HOMOLOGOUS structure is used to show that organisms share a common ancestor; it is when they have the same structure, but a different function. Ex: bird wing, whale fin, human arm.</li> <li>The wings of a moth, a bird, and a bat are examples of ANALOGOUS structures and are NOT evidence of a common ancestor; they have different structures, but the same function.</li> <li>A favorable characteristic passed on from parents to their offspring is called an ADAPTATION. An example is long necks on giraffes.</li> <li>Beneficial mutations that result in the survival of a species and result in an entirely different organisms over time is known as NATURAL SELECTION.</li> </ol>							
9.	BIOLOGICAL CLAS	SSIFICATION is a sy	stem used to organiz	e and show relati	onships between all organis	ms on Earth.		
10.	CHARLES DARWI	N is known as the f	father of evolution ar	d he studied the l	beaks of FINCHES.			
	8.L.1 - Structure	es and Functions of	f Living Things		8.L.5 - Molecular Biology	L		
פוס				SUNLIGHT	PHOTOSYNTHESIS	WATER		
		PATHOGEN	ΒΔCTFRIΔ			RGY/ATP		
F		RECT BAC	TERIA VIRUS					
	VECTOR	VACCINE	PARASITE					
				CALORI				
1.	A PATHOGEN cau	ises a disease; aka	a germ.					
2.	The four types of	pathogens are <b>BA</b>	CTERIA, VIRUS,	CARBOHYDRA	TES GLUCOSE	PROTEINS		
	FUNGI, and PARA	SITE.			BASAL METABOLIC RATE			
3.	A VIRUS is non-liv	ing and needs a he	ost cell to survive.	1 Dianta una th		a malea faad in		
4.	Examples are the A <b>FUNGUS</b> can re	flu and HIV/AIDS. produce both sexu	ally and asexually.	the form of <b>GLUCOSE</b> (sugar) through a process called				
_	An example is ath	lete's foot.		2 The reactants of photosynthesis are WATER				
5.	A BACTERIA can b	be treated by antib	piotics.	SUNUGHT and CARBON DIOXIDE				
6.	A PARASITE lives	on or in its host. A	n example is a tick	3. The products of photosynthesis are <b>GLUCOSE</b> and				
7	An <b>ANTIRIOTIC</b> is	used to treat bact	erial infections	OXYGEN.				
8.	A VACCINE is used	d to prevent viral i	nfections.	4. CELLULAR RESPIRATION breaks down the glucose				
9.	An <b>EPIDEMIC</b> is a	n outbreak of a dis	ease that occurs in	produced during photosynthesis to release a usable				
	one country or re	gion.		form of <b>ENERGY/ATP</b> .				
10.	<ol> <li>A PANDEMIC is an outbreak of a disease that affects a much larger region, usually global.</li> </ol>			LIPIDS, CARBOHYDRATES, and PROTEINS.				
11. The best way to prevent epidemics and pandemics is			7. If one consumes more calories than the body uses					
by everyone receiving VACCINATIONS.			then <b>THFY V</b>	VIII GAIN WFIGHT.	bouy uses,			
12. Pathogens can be transmitted by <b>DIRECT</b> contact such			8. A person wh	to burns calories at a fast rat	te is said to			
as contact with an infected person.			have a high	METABOLISM.				
such as a touching a doorknob or eating contaminated			9. BASAL METABOLIC RATE is the minimum number of					
food.				a person needs to maintair	ı bodily			
14. A <b>VECTOR</b> is a living organism that carries a pathogen functions v				nile at rest.				
<u>т</u> т.	from place to plac malaria.	ce. An example is a	a mosquito carrying					

		8.P.2 - Energy: Conservation and Transfer										
BIOMASS NONRENEWABLE RECYCLE WIND GLOBAL WARM				MING	NUCLEAR							
	OIL	FOSSIL FUELS	GEOTHER	MAL	REDUCE	GREENHOUSE	GASES	COAL				
		NATURAL GAS	SOLAR	HYDRO	ELECTRIC	RENEWABLE	REUSE					
1.	1. Energy resources that can be replaced or reused are called <b>RENEWABLE</b> energy resources.											
2.	Examples of renewable energy resources are: WIND (energy from moving air), SOLAR (energy from the sun), HYDROELECTRIC (energy from moving water), BIOMASS (energy from plant and animal material), and GEOTHERMAL											
3.	(energy from earth's heat). Energy resources that cannot be replaced or are used at a faster rate than which they are formed at are called											
4.	NONRENEWABLE energy resources. Examples of nonrenewable energy resources are: FOSSIL FUELS and NUCLEAR energy.											
5.	The three	e types of fossil fuels ar	e: COAL, OIL, a	nd NATU	RAL GAS.							
6.	Burning f	ossil fuels releases GRE	ENHOUSE GAS	SES into t	he atmosphere,	which can lead to GI	LOBAL WAR	MING.				
7.	REDUCE,	<b>REUSE</b> and <b>RECYCLE</b> ar	e three ways t	hat huma	ns can conserve	natural resources.						
			<u>8.E.1 - Earth</u>	Systems,	Structures, and	d Processes						
	SUNLIGH	IT NEUTRAL	VARIET	Y	FROZEN	<b>RIVER BASIN</b>	DEPTH	3%				
I	JPWELLIN	G BIOINDICA	TORS	HIGH	POINT	TEMPERATU	RE	TURBIDITY				
	OXYGEN	POTABLE	AQUIFERS		NONPOINT	TREATED	COLD	PH				
	NITE	RATES PRESSUR	E EST	UARY	97%	GLACIERS	TEMPERAT	URE				
<ol> <li>The Earth is 97% saltwater and 3% freshwater.</li> <li>Most of the freshwater on Earth is FROZEN in polar ice caps and GLACIERS</li> <li>A RIVER BASIN is the land drained by a river and its tributaries.</li> <li>UPWELLING occurs in an area where nutrients are drawn to the surface of the ocean by the movement of cold water from the bottom of the ocean to the top. Hint: there would be a lot of fish here!</li> <li>Life in the occan is determined by 3 factors: SUNLIGHT, TEMPERATURE, and DEPTH.</li> <li>As you move from the open ocean to the deep ocean the amount of sunlight decrease, the TEMPERATURE decrease, and the PRESSURE increases.</li> <li>An ESTUARY is where fresh and saltwater mix. These areas have a wide variety of life and are more protected than the open ocean.</li> <li>Dissolved OXYGEN is a measure of the amount of oxygen in water that is available for aquatic organisms. There is more dissolved oxygen in COLD water.</li> <li>PH is a measure of how acidic or basic water is. When this measure of water quality is 7, the water is NEUTRAL.</li> <li>NITRATES are nutrients for plants that most often get into our water from agricultural runoff.</li> <li>TURBIDITY is a measure of how clear water is.</li> <li>HIGH turbidity can lead to lack of sunlight and is an indication of unhealthy water.</li> <li>BIOINDICATORS are living organisms in a body of water that are sensitive to pollution. A wide VARIETY of these organisms indicates healthy water.</li> <li>AQUIFERS are large deposits of groundwater in rock layers that can be extracted and used.</li> <li>POINT source pollution comes from a single known source. For example: a factory is dumping nuclear waste into a river.</li> </ol>												
l	chemicals.											
			G	00	D LUCK	<b>(!</b>		GOOD LUCK!				

## YOU HAVE WORKED HARD...TIME TO SHINE!

Big Goal: <u>ALL students</u> earn a <u>4 or 5</u> on the 8<sup>th</sup> grade science EOG!